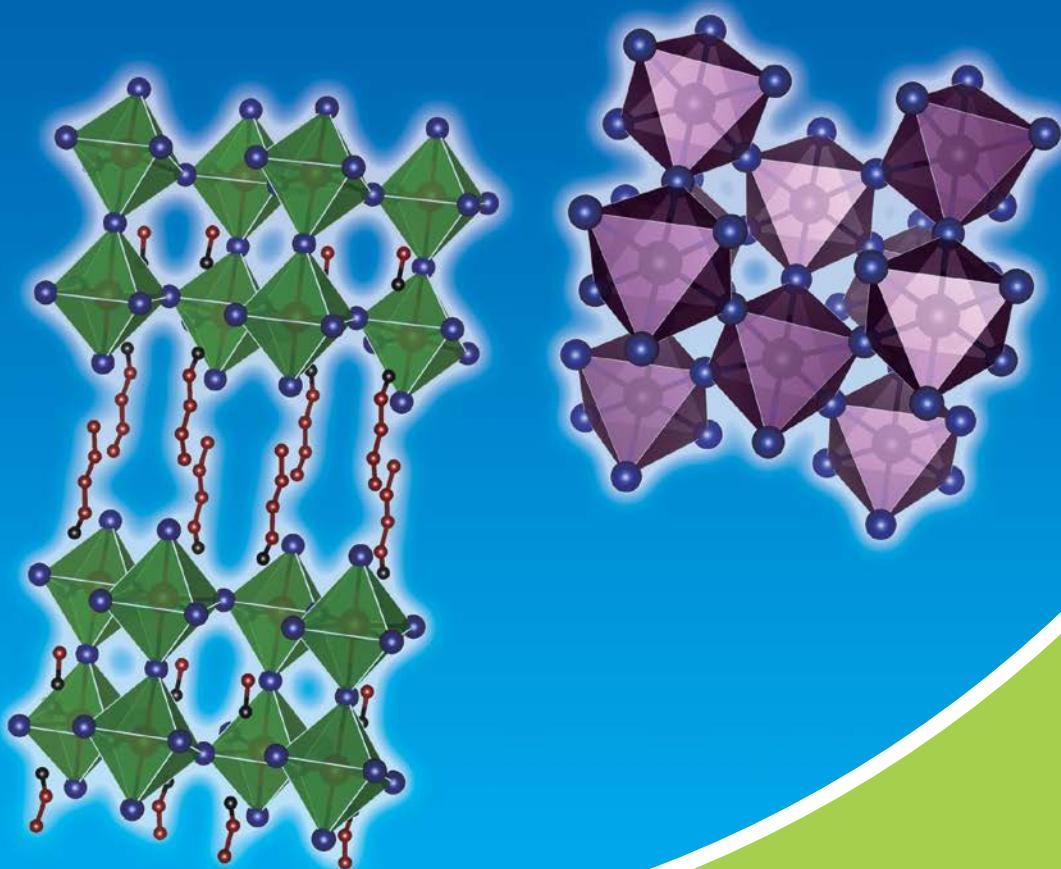


Organic-Inorganic Perovskite Precursors



Organic Onium Salts

Lead Halides

Other Lead Compounds

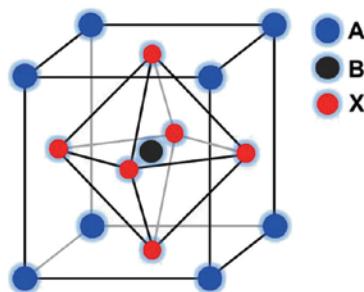
Cesium Halides

Bismuth Halides

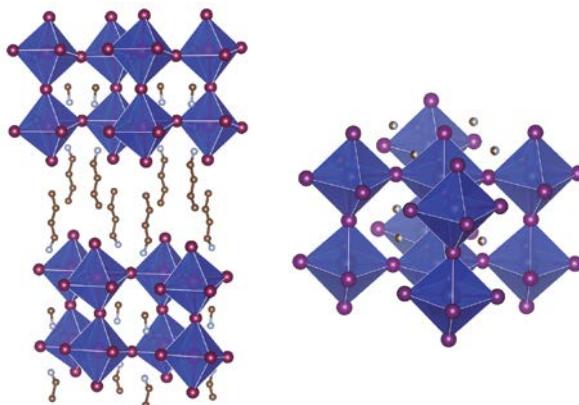
Tin Halides

Organic-Inorganic Perovskite Precursors

"Perovskite" originates from the mineral name of calcium titanate (CaTiO_3) and the compounds with formula of ABX_3 generally belong to a perovskite-type compound, where the A is a divalent and B is a tetravalent metal ion. A perovskite with cubic or orthorhombic phases shows ferroelectricity, for instance, barium titanate (BaTiO_3) is a ferroelectric or piezoelectric material.¹⁾ High temperature superconductive oxides with a unit of copper oxide are obtained from all perovskite compounds.²⁾ These perovskite compounds consist of metal ions and oxygen atoms, and are manufactured by a physical procedure (eg. sintering method).³⁾ Modification of the metal ion and a changing ratio of the metal ion components can drastically control physical properties of the perovskite. In addition to the oxide perovskites, halide-based perovskites are also well known.



On the other hand, one can replace the cationic component with an organic ammonium at the A site. In this case, a chemical method can provide a perovskite compound. This perovskite compound is called an "organic-inorganic perovskite compound", because it contains an organic component. A metal ion component usually involves tin or lead.^{4,5)} This perovskite compound has the general formula $[(\text{RNH}_3)_m\text{MX}_n]$, in which modifications of metal (M), halide (X) and organic groups (R) precisely control physical properties. Among them, the tin perovskite is relatively better for electrical conduction,⁶⁾ and the lead one is better for optical properties.⁷⁾ A chemical modification of the halide controls band gap.⁸⁾ Selection of organic onium halide, metal halide and their mixing ratio changes the component ratio of the halide. The organic groups are selected from methyl, long alkyls, phenyl, benzyl, phenethyl and so on. Diversity of these organic groups allows controlling the structure of a perovskite compound. For instance, a perovskite compound with R = methyl provides $[(\text{MeNH}_3)\text{MX}_3]$ having a three-dimensional cubic perovskite structure.⁹⁾ A perovskite compound with R = $\text{C}_n\text{H}_{2n+1}$ ($n \geq 2$) provides a two-dimensional perovskite layer and the length of alkyl group can control the inter-layer distance.¹⁰⁾



An application of an organic-inorganic perovskite is a perovskite solar cell.¹¹⁻¹⁵⁾ This solar cell can usually be fabricated by the three-dimensional cubic perovskite $[(\text{MeNH}_3)\text{MX}_3]$. Doping effects of formamidinium¹⁶⁾ and cesium cations¹⁷⁾ to the A site were also investigated for the perovskite solar cell research. Research on the perovskite solar cell recently received much attention. Power conversion efficiency of this solar cell is more than those of organic photovoltaics (OPV) and dye-sensitized solar cells (DSSC), and the device can be fabricated by a solution method at low cost.

References

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Organic Onium Salts

Chloride Salts

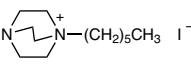
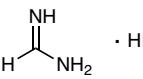
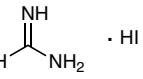
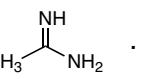
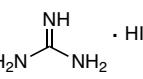
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Bromide Salts													

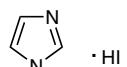
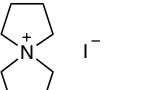
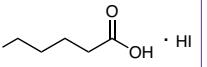
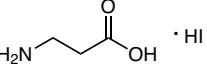
Organic-Inorganic Perovskite Precursors

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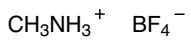
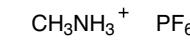
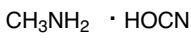
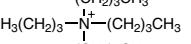
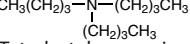
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P2212 CH ₃ CH ₂ CH ₂ NH ₂ · HI Propylamine Hydroiodide CAS RN: 14488-45-0	B4433 CH ₃ (CH ₂) ₃ NH ₂ · HI Butylamine Hydroiodide CAS RN: 36945-08-1	I0935 CH ₃ CH(CH ₃) ₂ NH ₂ · HI Isobutylamine Hydroiodide CAS RN: 205508-75-4	B4434 CH ₃ C(CH ₃) ₃ NH ₂ · HI tert-Butylamine Hydroiodide CAS RN: 39557-45-4	P2740 CH ₃ CH ₂ CH ₂ CH ₂ CH ₂ NH ₂ · HI Pentylamine Hydroiodide CAS RN: 60762-85-8	
I1095 CH ₃ CH ₂ CH(CH ₃) ₂ NH ₂ · HI Isopentylamine Hydroiodide	N1157 CH ₃ C(CH ₃) ₃ CH ₂ NH ₂ · HI Neopentylamine Hydroiodide	O0485 CH ₃ CH ₂ NH ₂ · HI n-Octylammonium Iodide CAS RN: 60734-63-6	T3785 CH ₃ CH ₂ CH(CH ₃) ₂ CH ₂ C(CH ₃) ₃ NH ₂ · HI tert-Octylamine Hydroiodide	D5538 CH ₃ (CH ₂) ₁₁ NH ₂ · HI Dodecylamine Hydroiodide CAS RN: 34099-97-3	
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Organic-Inorganic Perovskite Precursors

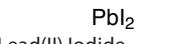
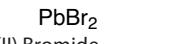
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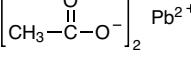
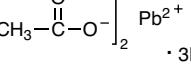
Pseudo Halide Salts

M2990	1g 5g	M2989	1g 5g	M3134	1g 5g	T0914	25g 100g 500g	T2648	25g
 Methylamine Tetrafluoroborate CAS RN: 42539-74-2		 Methylammonium Hexafluorophosphate CAS RN: 28302-50-3		 Methylamine Cyanate CAS RN: 63405-91-4		 Tetrabutylammonium Tetrafluoroborate CAS RN: 429-42-5		 Tetrabutylammonium Tetrafluoroborate (Br <0.02 %) CAS RN: 429-42-5	

Lead Halides

L0279	1g 5g 25g 100g 1kg	L0288	1g 5g 25g	L0346	1g 5g
	 Lead(II) Iodide (99.99%, trace metals basis) [for Perovskite precursor] CAS RN: 10101-63-0		 Lead(II) Bromide [for Perovskite precursor] CAS RN: 10031-22-8		 Lead(II) Bromide (Low water content) [for Perovskite precursor] CAS RN: 10031-22-8

L0291	1g 5g	L0292	1g 5g 25g	C3569	1g 5g	C3570	1g 5g
	 Lead(II) Chloride (purified by sublimation) [for Perovskite precursor] CAS RN: 7758-95-4		 Lead(II) Chloride [for Perovskite precursor] CAS RN: 7758-95-4		 CsPbBr ₃		 CsPbI ₃

L0315	1g 5g 25g	L0330	25g 100g
	 Lead(II) Acetate [for Perovskite precursor] CAS RN: 301-04-2		 Lead(II) Acetate Trihydrate CAS RN: 6080-56-4

C2202	25g 100g	C2203	25g 100g	C2205	25g
	 Cesium Bromide CAS RN: 7787-69-1		 Cesium Chloride CAS RN: 7647-17-8		 Cesium Iodide CAS RN: 7789-17-5

Bismuth Halides

B5787	5g 25g
<chem>BiI3</chem>	
Bismuth(III) Iodide Anhydrous CAS RN: 7787-64-6	

Tin Halides

T3449	1g 5g	T3570	1g 5g	T3573	1g 5g
<chem>SnI2</chem>		<chem>SnCl2</chem>		<chem>SnBr2</chem>	
Tin(II) Iodide [for Perovskite precursor] CAS RN: 10294-70-9		Tin(II) Chloride [for Perovskite precursor] CAS RN: 7772-99-8		Tin(II) Bromide [for Perovskite precursor] CAS RN: 10031-24-0	

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